

# CW SRF Systems with Ingot Niobium\* and their Applications

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**ISOHIM/University of Virginia/JLab**

**\*CBMM – JLab Technology**

# Overview

- ✓ **Brief introduction to the largest SRF CW Accelerator**
- ✓ **Fine grain and ingot niobium technologies**
- ✓ **Qo Improvement Program**
- ✓ **Ingot niobium CW Applications**

# Glossary 1

- Niobium – highly ductile refractory metal with highest superconducting transition temperature ( $T_c \sim 9.25 \text{ K}$ ) at which the electrical resistance drops to Zero, highest peak magnetic field  $H_{pk}$  & high melting temperature  $T_m$
- RRR – Residual Resistance Ratio  $\sim R_{300}/R_{4.2}$
- Important Interstitials H, C, N and O that contribute to RRR significantly
  - and tantalum, substitutional impurity does not significantly contribute
- Surface Resistance  $R_s = R_{BCS} + R_0$   
 $R_{BCS}$  depends on surface magnetic field, temperature and frequency

# Glossary 2

- Quality Factor  $Q_0 = G/R_s$ , where  $G$  is the geometry factor and it is independent of the cavity frequency (ideal  $\sim 2 \times 10^{11}$ )
- $H_{pk}$  Surface peak magnetic field (mT)
- $E_{acc}$  Accelerating gradient (MV/m)
- Optimized Processes and procedures: forming cells, cleaning, welding, surface treatments and stress relieving processes and final contamination free evacuation

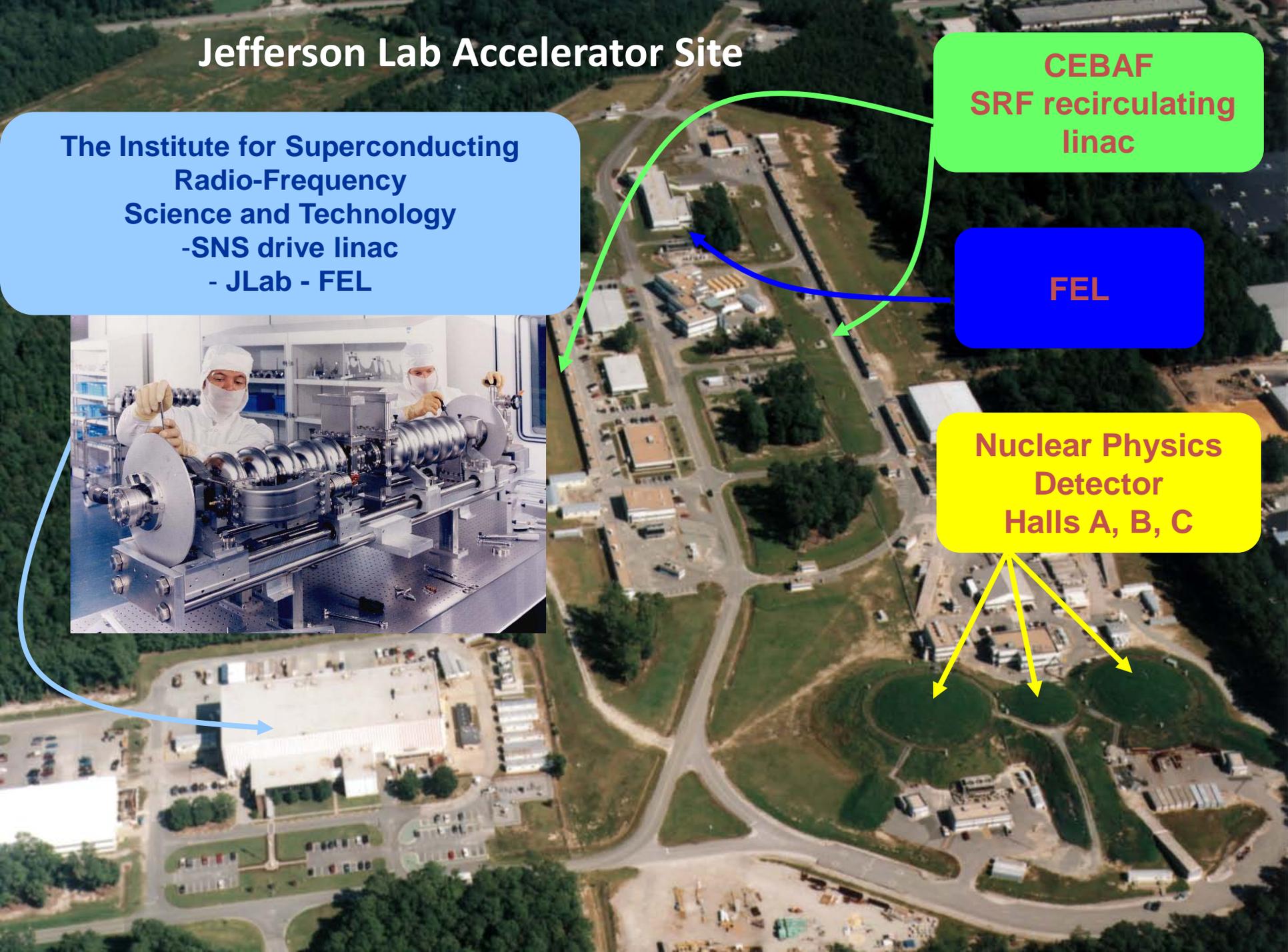
# Jefferson Lab Accelerator Site

The Institute for Superconducting  
Radio-Frequency  
Science and Technology  
-SNS drive linac  
- JLab - FEL

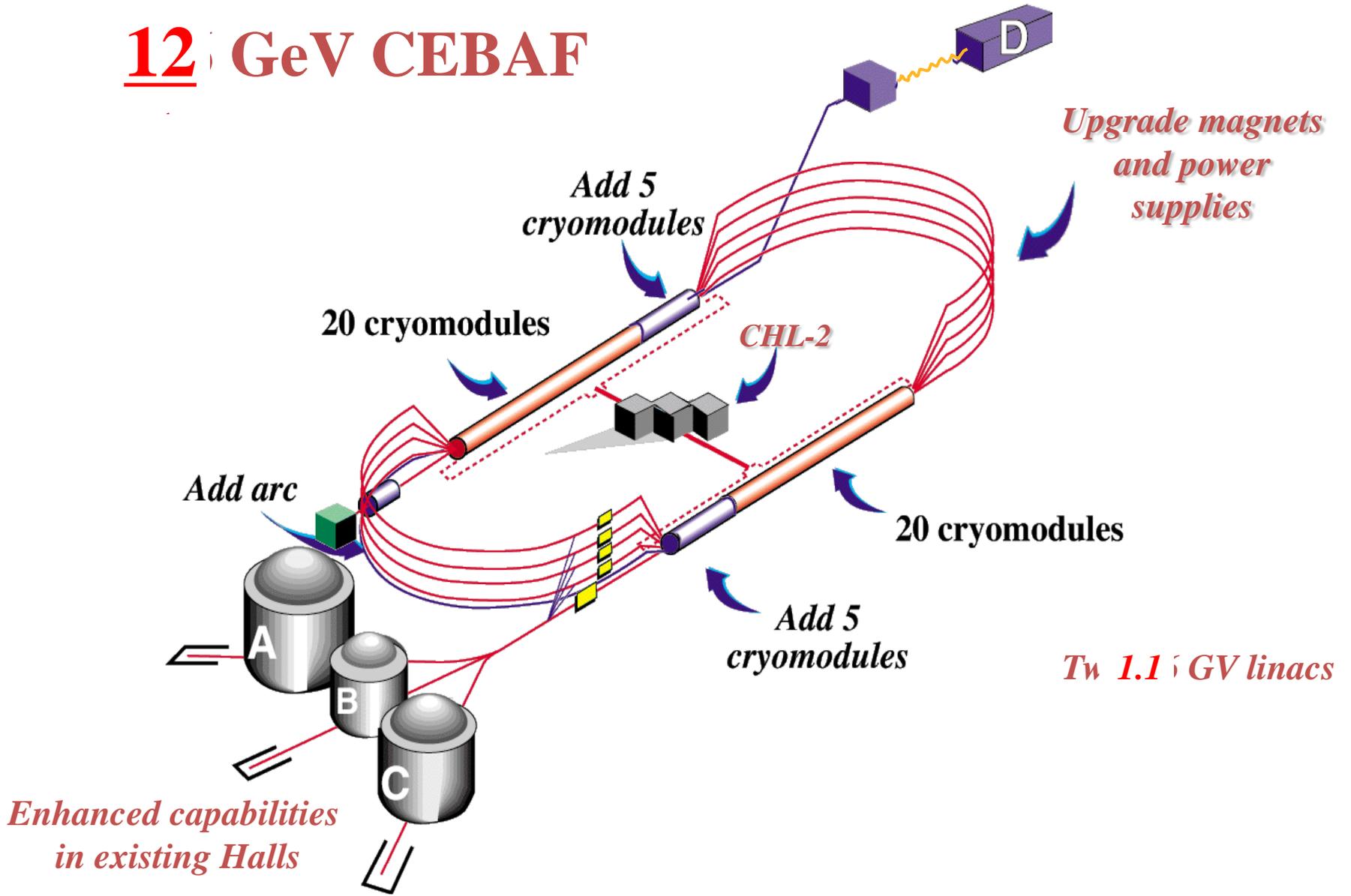
CEBAF  
SRF recirculating  
linac

FEL

Nuclear Physics  
Detector  
Halls A, B, C



# 12 GeV CEBAF



# Highlights of Early SRF Technology

- Cavities were mostly made from ingot niobium
  - Process and procedures were similar and as varied as today
- Reactor grade Niobium material in ingot, bar, plate sheet and tube form was available
- Achievable gradient limited by multipacting and/or field emission
- Residual surface resistance ( $n\Omega$ ) was not well understood
  - Still the case
- At highest frequencies (Electropolished fine grain, X-band)  $H_{pk} \sim 159$  mT  $Q_0 \sim 5 \times 10^9$
- (BCP'd ingot Nb, 1970's)  $H_{pk} \sim 108$  mT &  $Q_0 \sim 1 \times 10^{11}$  @ 1.2 K CW
- For comparison (CEBAF upgrade spec.)  $H_{pk} \sim 76$  mT  $Q_0 \sim 7 \times 10^9$  @ 2 K CW (2008)

# Historical Example of Ingot Niobium 1

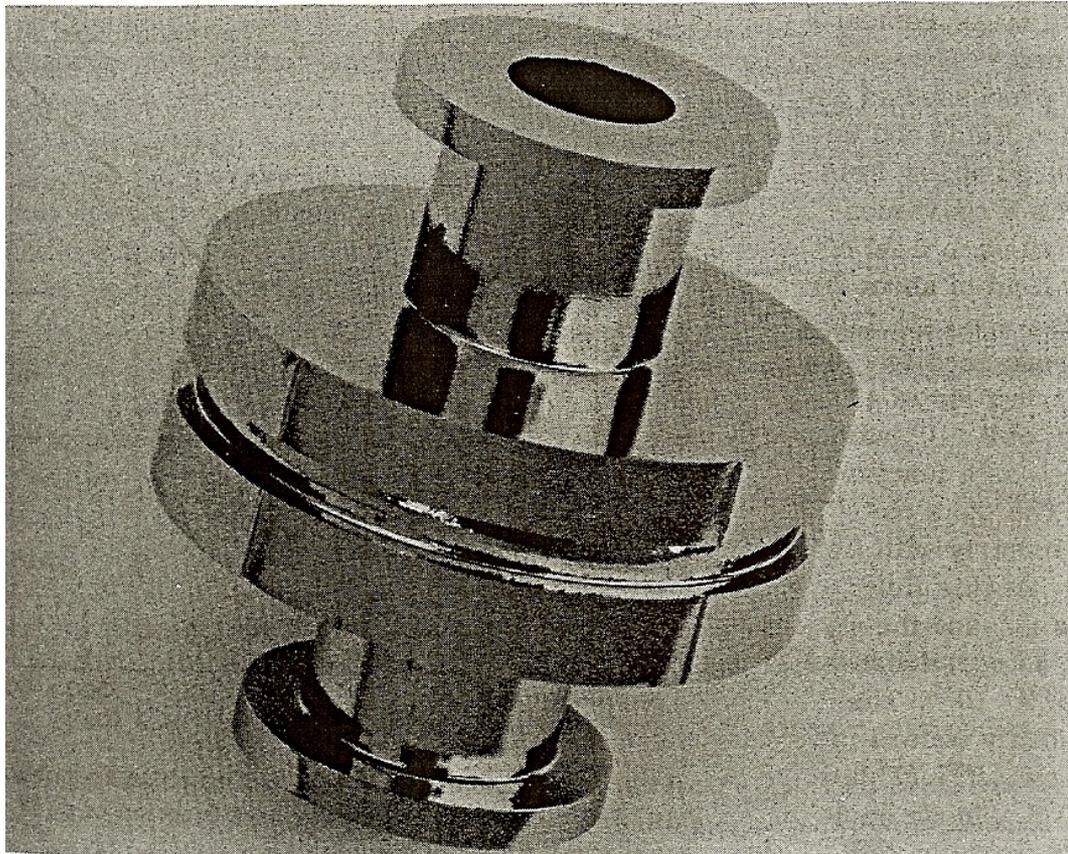


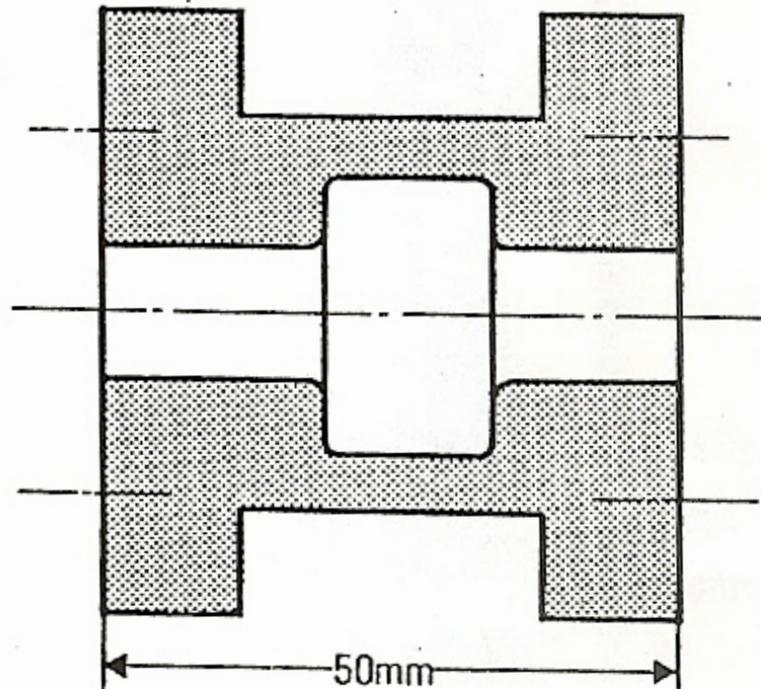
FIG. 1. An electron-beam welded  $TM_{010}$  mode Nb cavity. The cavity is resonant at 8.6 GHz and is 3.6 cm in overall length.

$H_{pk} \sim 108$  mT with BCP

Stanford solid niobium cavity 1970

# Historical Example of Ingot Niobium 2

## Siemens solid niobium cavity 1973



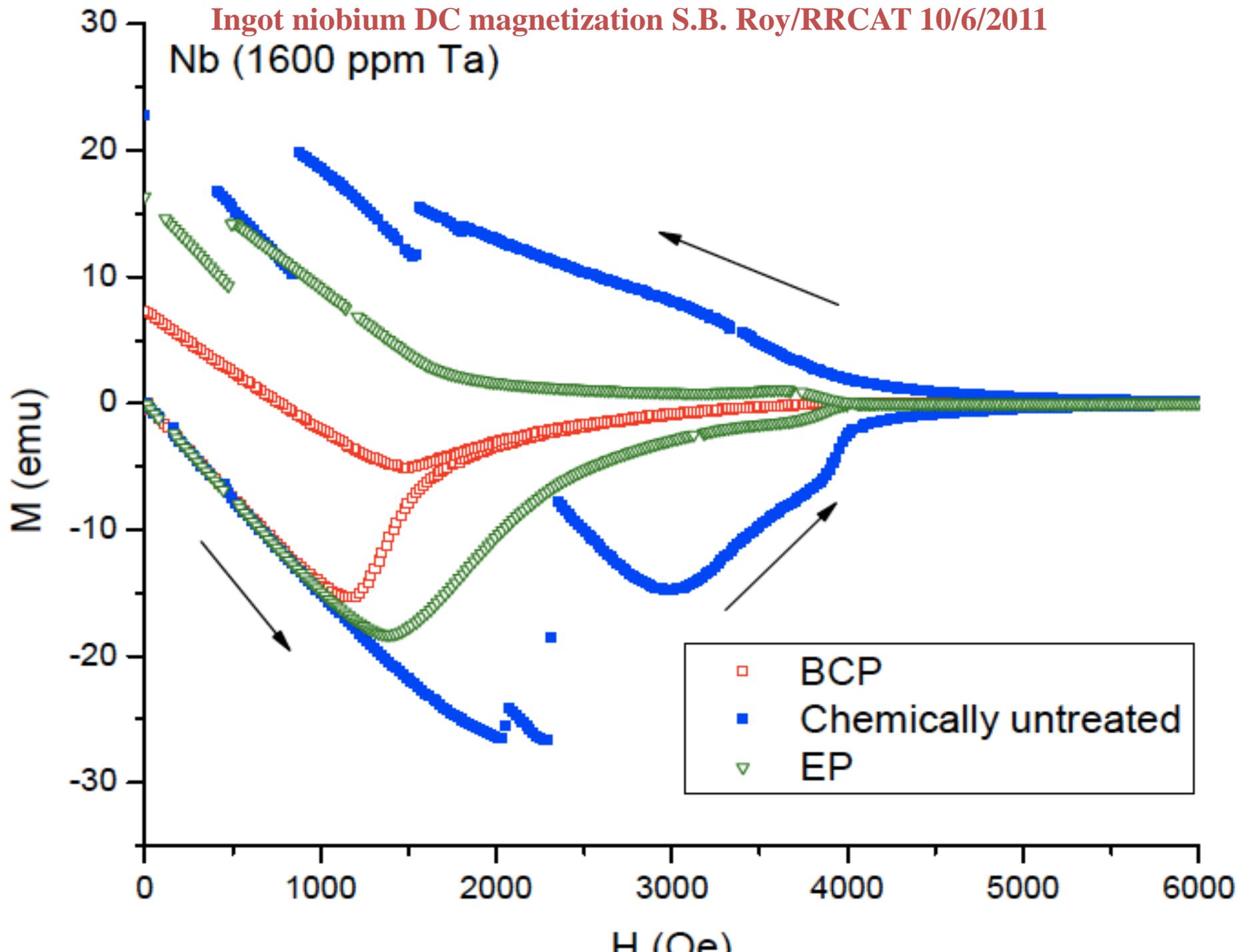
$H_{pk} \sim 109$  mT with BCP

$H_{pk} \sim 130$  mT with EP

Fig. 1. Single piece  $TM_{010}$ -niobium cavity with a resonant frequency of 9.5 GHz.

**EP'd reactor grade fine grain niobium cavity set a record  $H_{pk}$  of 159 mT**

Nb (1600 ppm Ta)



# Multi cell cavity fabrication

Forming



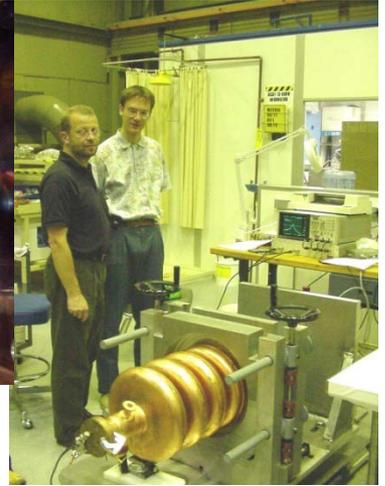
Machining



Welding

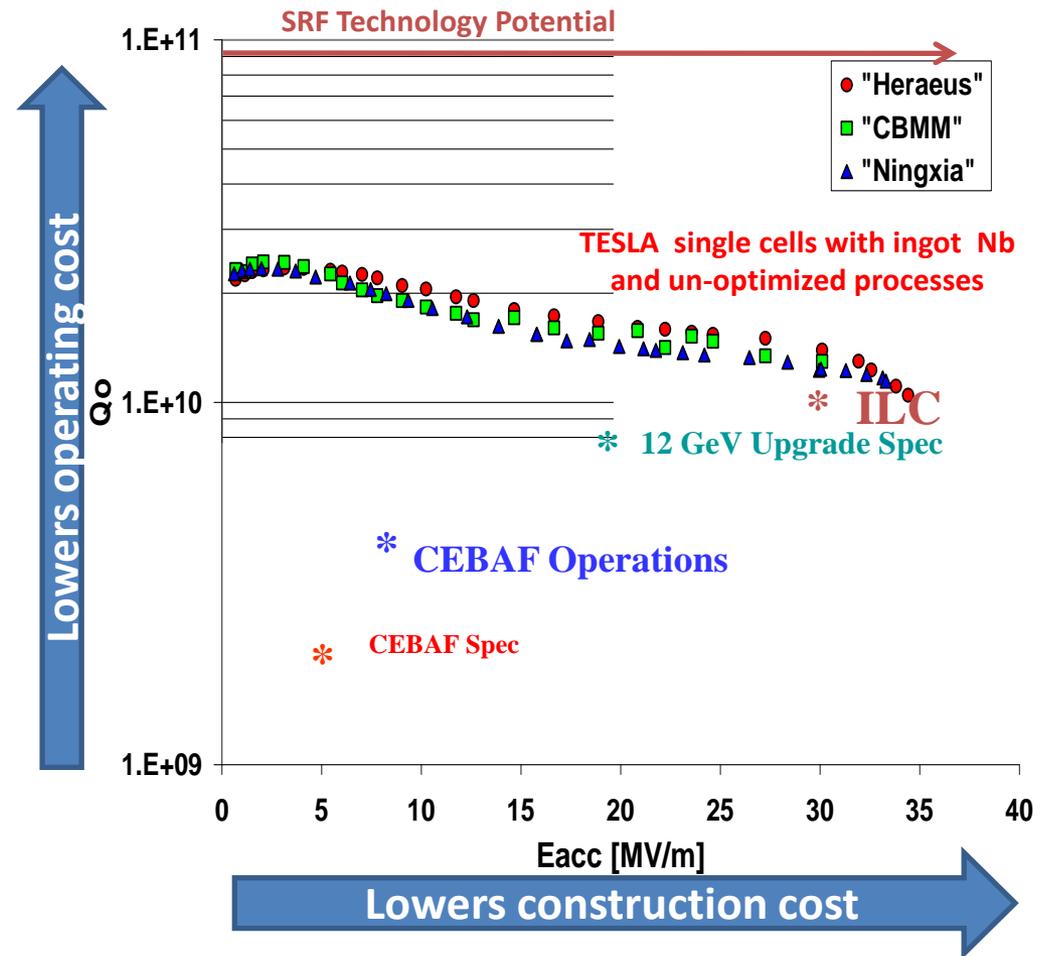


Tuning



**>80% of CEBAF cavities were made with CBMM Pyrochlore ore based niobium  
In comparison to present day use of Tantalite/Columbite ore based niobium**

# Niobium cavity – performance (CW)



In nearly 40 years  $E_{acc}$  improved by a factor of 5, now DOE NP and JLab working to improve  $Q_0$  by a factor of ~3

# Comparison of fine grain and ingot niobium

# Niobium Specifications – Past & Present (@JLab)

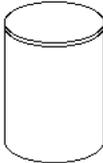
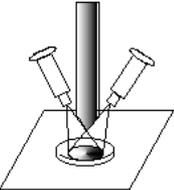
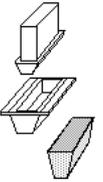
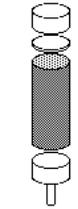
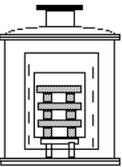
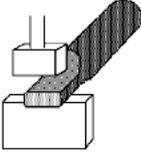
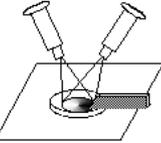
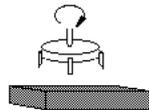
- Polycrystalline Niobium with ASTM #5 Grain Size or finer ~ 50 micro meters & 90% recrystallized
- Percentage of elongation > 25
- Yield Strength > 10.7 KSI (~75 MPa) (7 KSI for SNS)
- RRR > 250
- Tantalum < 1000 wt ppm

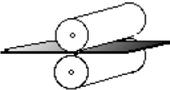
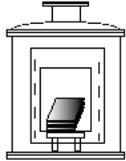
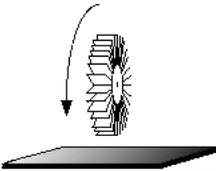
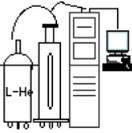
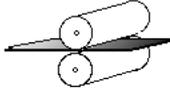
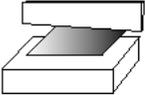
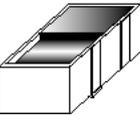
**Note:**

Recrystallization and high yield strength (YS) are mutually exclusive, the “kiss pass” used for increasing the YS introduces significant surface damage

**Note: These specifications are wrt the physical structure only & do not include SC properties**

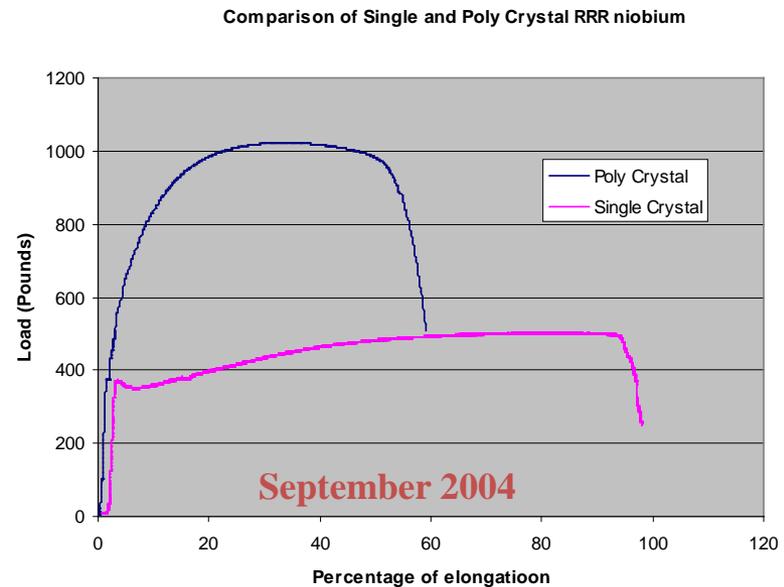
# Process steps - fine grain Niobium

Fabrication process of Nb sheets for Superconducting Cavities	
Tokyo Denkai Co., Ltd. <span style="float: right;">H.Umezawa</span>	
1. Mother Material 	5. EB Melting (2nd, 3rd) 
2. Pressing 	6. Cutting 
3. Out gassing and Sintering 	7. Forging 
4. EB Melting(1st) 	8. Mechanical grinding 

9. Rolling 	13. Annealing 
10. Polishing 	14. Testing 
11. Rolling 	15. Polishing 
12. Cutting 	16. Packing 

Note **During this process foreign materials can be embedded so QA is required**

# Birth of Ingot Niobium Technology CBMM-JLab CRADA, August 2004



**Chosen for Excellent Ductility and Surface Smoothness with just BCP**

**First CBMM/JLab International Patents were applied for in April, 2005**

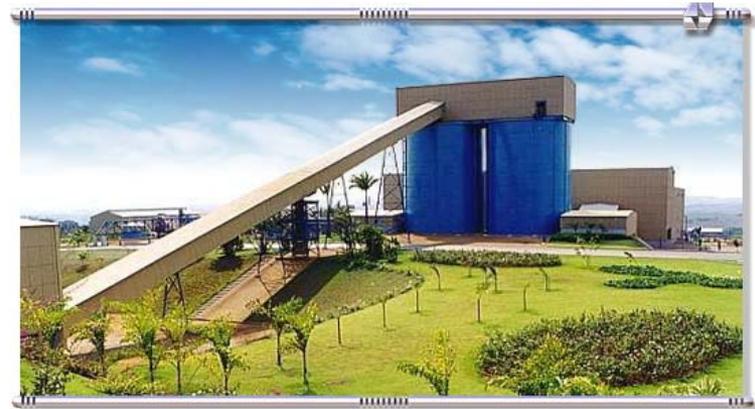
# Araxá Mine in Brazil & RRR Niobium

From ore to oxide to large grain ingots

The CBMM open cast mine



Conveyor belt bringing the ore to concentration plant

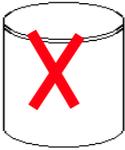
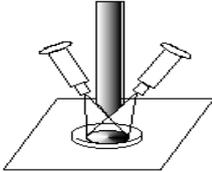
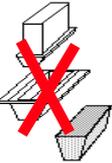
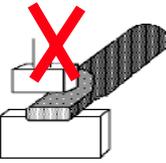
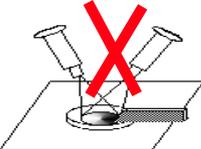


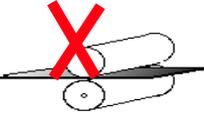
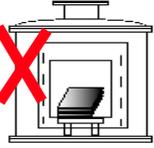
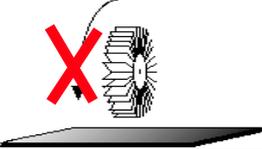
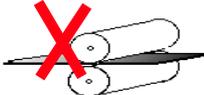
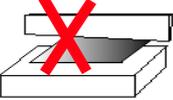
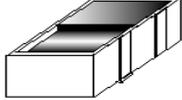
Electron beam furnace for the refinement of Niobium metal, producing 210 tonnes per annum



Finished RRR Nb ingot from the Pyrochlore ore

# Economic path for CW applications

Fabrication process of Nb sheets for Superconducting Cavities	
Tokyo Denkai Co., Ltd. <span style="float: right;">H.Umezawa</span>	
<p>1. Mother Material</p> 	<p>5. EB Melting (2nd, 3rd)</p> 
<p>2. Pressing</p> 	<p>6. Cutting</p> 
<p>3. Out gassing and Sintering</p> 	<p>7. Forging</p> 
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<p>9. Rolling</p> 	<p>13. Annealing</p> 
<p>10. Polishing</p> 	<p>14. Testing</p> 
<p>11. Rolling</p> 	<p>15. Polishing</p> 
<p>12. Cutting</p> 	<p>16. Packing</p> 
<p>Note: <b>Cost of the ingot sliced Nb sheets anticipated to be less than half of polycrystalline Nb &amp; no QA</b></p>	

# Extrinsic and intrinsic contamination of Nb determines the performance of the cavities

## Extrinsic

- Surface contamination
  - Molecular and particulate

## Intrinsic

- Niobium is a prolific hydrogen absorber in the absence of the natural surface oxide
  - Hydride formation

# Vacuum Contamination Work Shop at JLab 1997



**Minimizing organic and particulate recontamination addressed**

**Re-contamination prevention courses were organized at JLab in 2000 and 2005**

# **International Symposium On Hydrogen In Matter (ISOHIM) Publications**

**Hydrogen in Materials and Vacuum Systems AIP CP 671**

<http://www.virtualjournals.org/dbt/dbt.jsp?KEY=APCPCS&Volume=671&Issue=1>

**Hydrogen in Matter AIP CP 837**

<http://www.virtualjournals.org/dbt/dbt.jsp?KEY=APCPCS&Volume=837&Issue=1>

**Single Crystal Large Grain Niobium AIP CP 927**

<http://www.virtualjournals.org/dbt/dbt.jsp?KEY=APCPCS&Volume=927&Issue=1>

**Superconducting Science and Technology of Ingot Niobium AIP CP 1352**

<http://scitation.aip.org/dbt/dbt.jsp?KEY=APCPCS&Volume=1352&Issue=1>

# **Q<sub>0</sub>, hydrogen & cavity performance**

- **Q-disease in the cavities is an example of a gross manifestation of hydrogen effect similar to gross air leak in high vacuum systems**
- **As we are looking to improve the cavity performance (Q) further we need to understand the effects of proton in niobium and take steps to minimize the solid-solution of protons, similar to eliminating smaller air leaks in UHV systems**
- **Hydrogen is difficult to measure quantitatively at the concentration levels that we have to in materials in general and greatly in niobium**
- **Like vacuum leak standards, we need to develop Nb-hydrogen standards**

# Goals of the Qo Improvement Program

## 1. Scientific Understanding

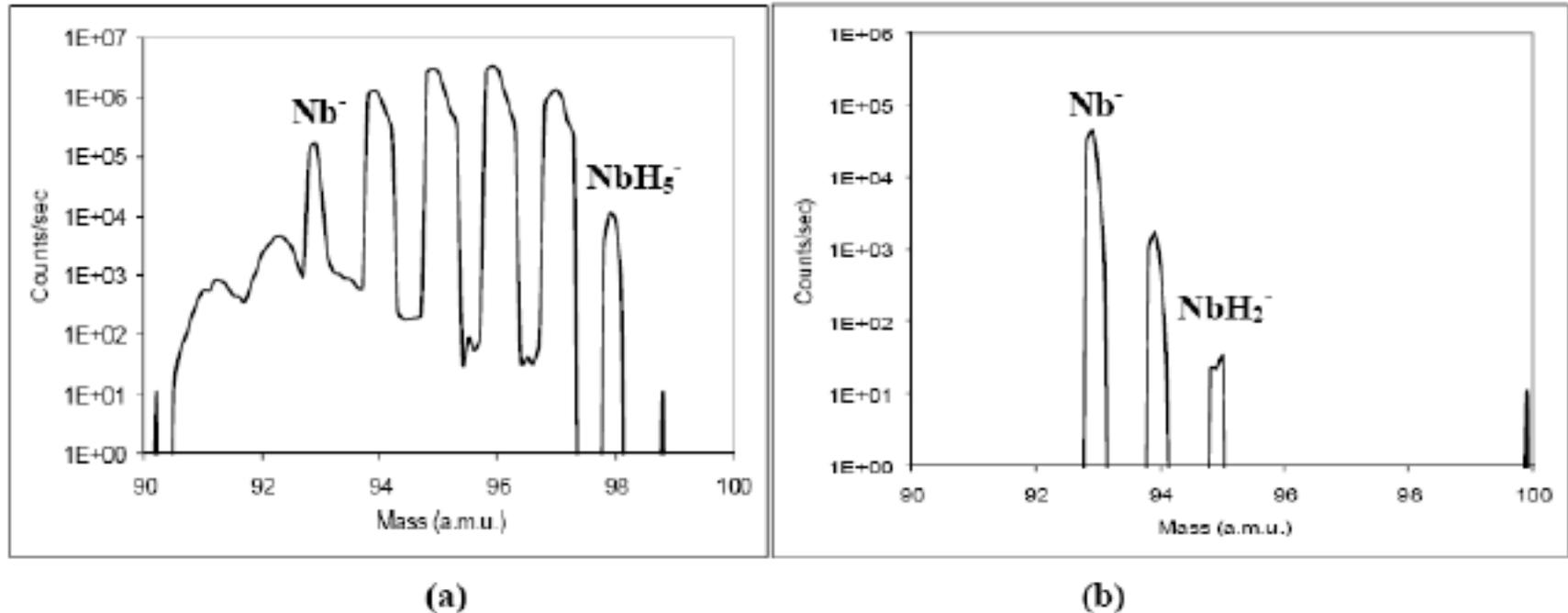
# Hydrogen absorption with BCP and EP

- **Very high equilibrium hydrogen activities (fugacity) have been estimated when Nb metal is in contact with water or BCP solution**
- **Hydrogen is readily absorbed into Nb when the protective oxide layer is removed**
- **Lower H fugacity's are obtained due to an anodic polarization of Nb during EP and hence lower hydrogen absorption**

**R.E. Ricker, G. R. Myneni, J. Res. Natl. Inst. Stand. Technol. 115, 353-371 (2010)**

**NIST/JLab**

## High temperature annealing removes gross hydrogen



**FIGURE 1.** SIMS mass spectra showing difference in H between (a) non-heat treated and (b) heat treated sample.

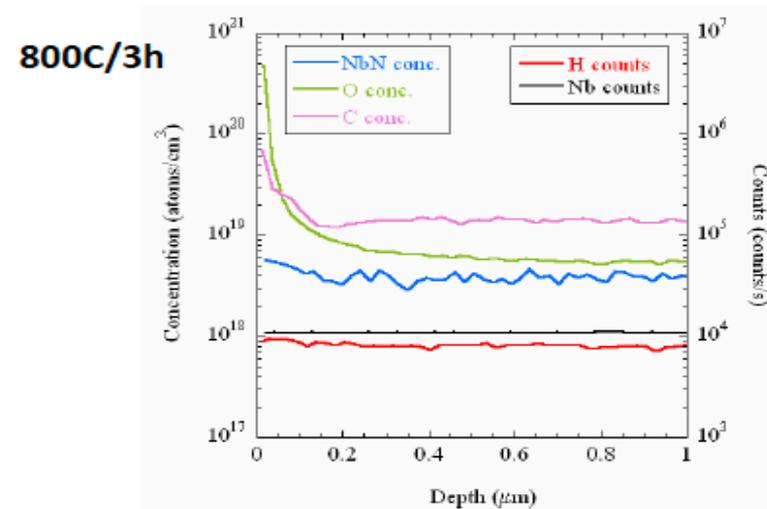
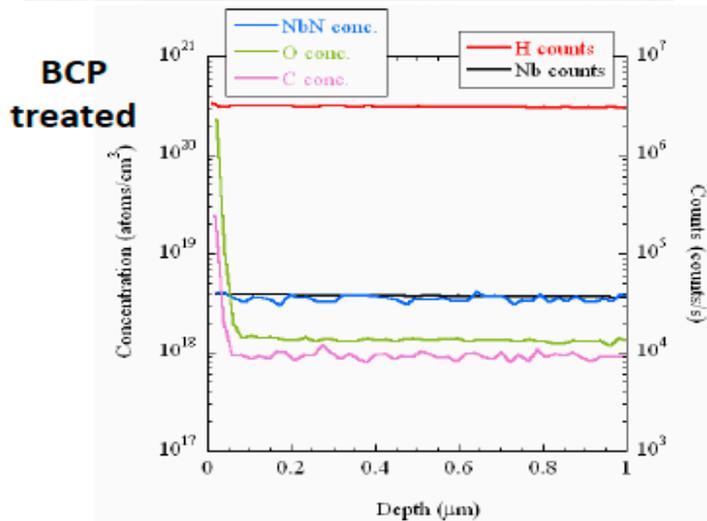
**NbH (Beta Phase) very much in existence after anneals**

# Heat treatment to remove hydrogen



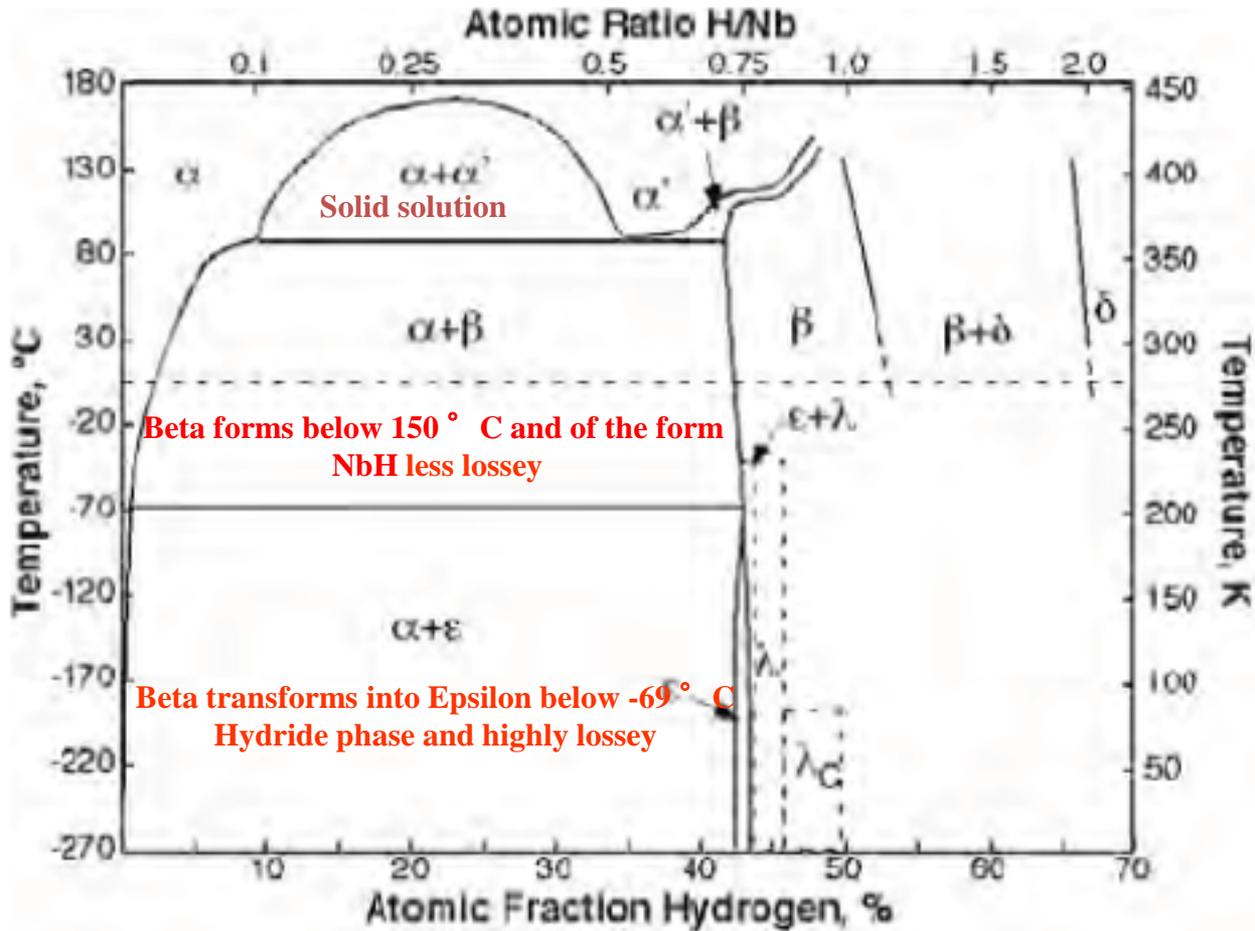
- 800°C/3h, pressure  $\sim 10^{-6}$  mbar
- **No chemical etching afterwards!**
- Nb samples were treated with the cavities and depth profiling of the impurities was done at NCSU

**$\sim 2$  orders of magnitude lower hydrogen content after HT**

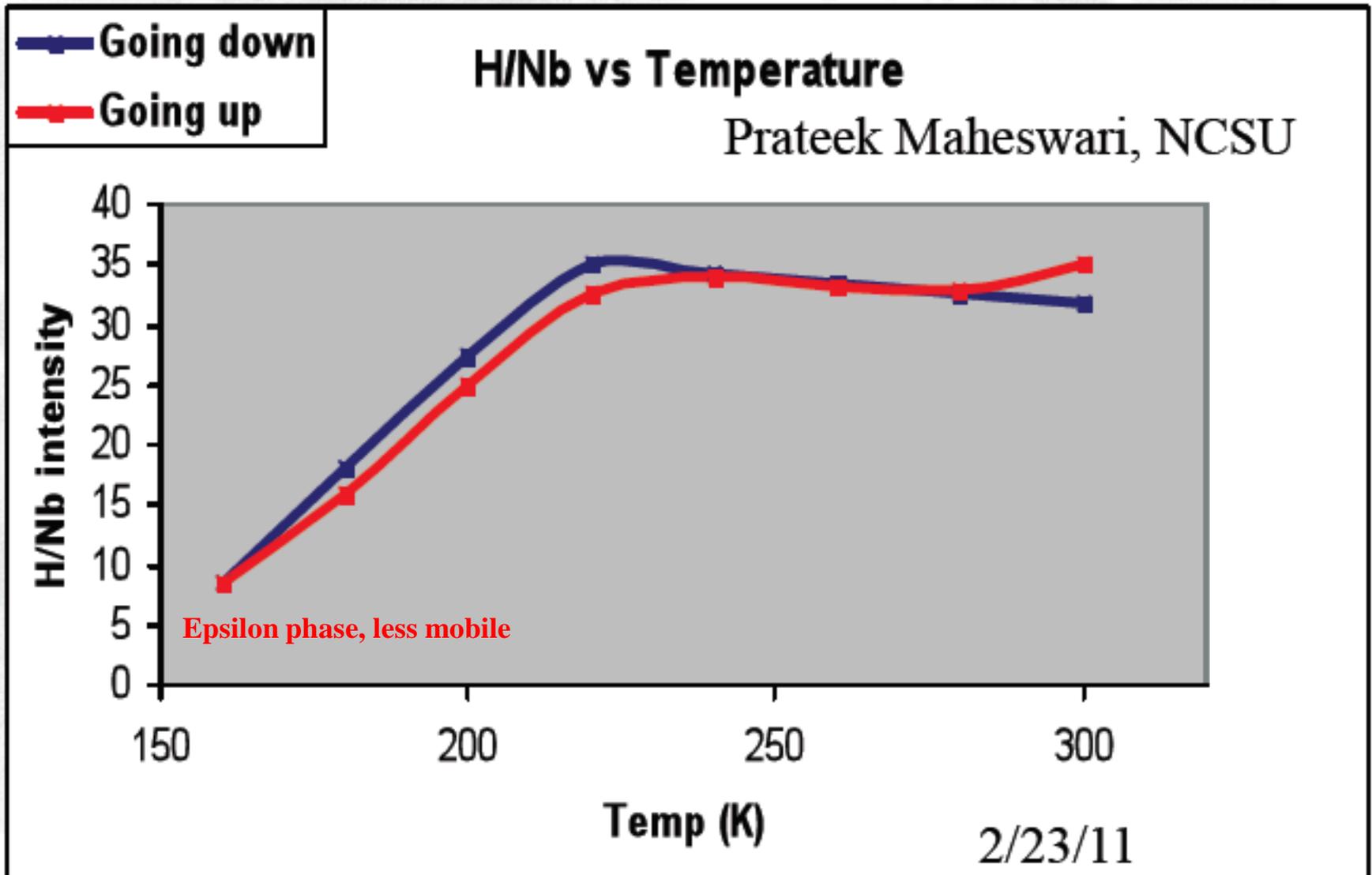


**Currently used furnaces contaminate the cavity surfaces, chemical re-etching reintroduces H**

# Niobium – hydrogen phase diagram



# Hydrogen phase change

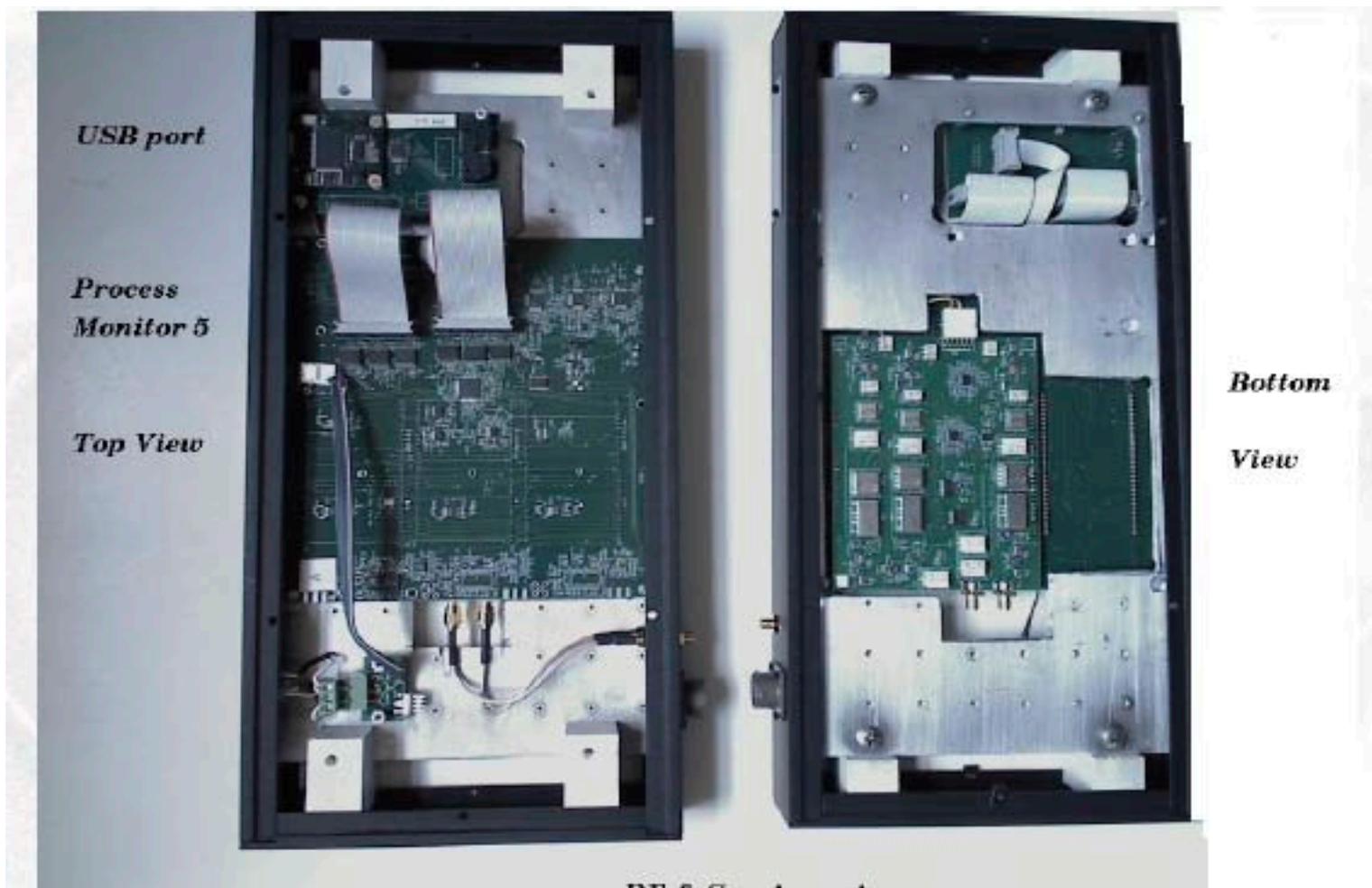




# Goals of the Qo Improvement Program

## 2. Technology Development

# Eddy current (0.1 to 2 GHz) & Optical measurement system

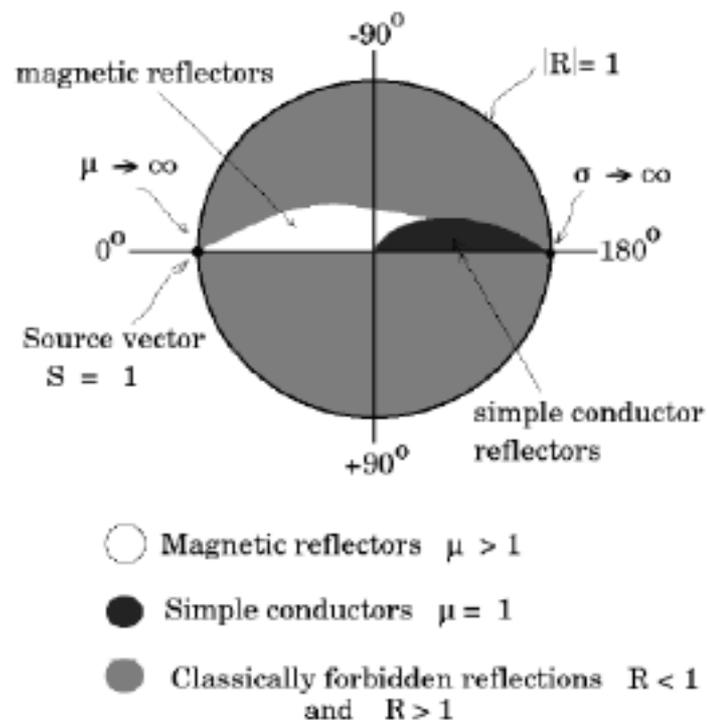
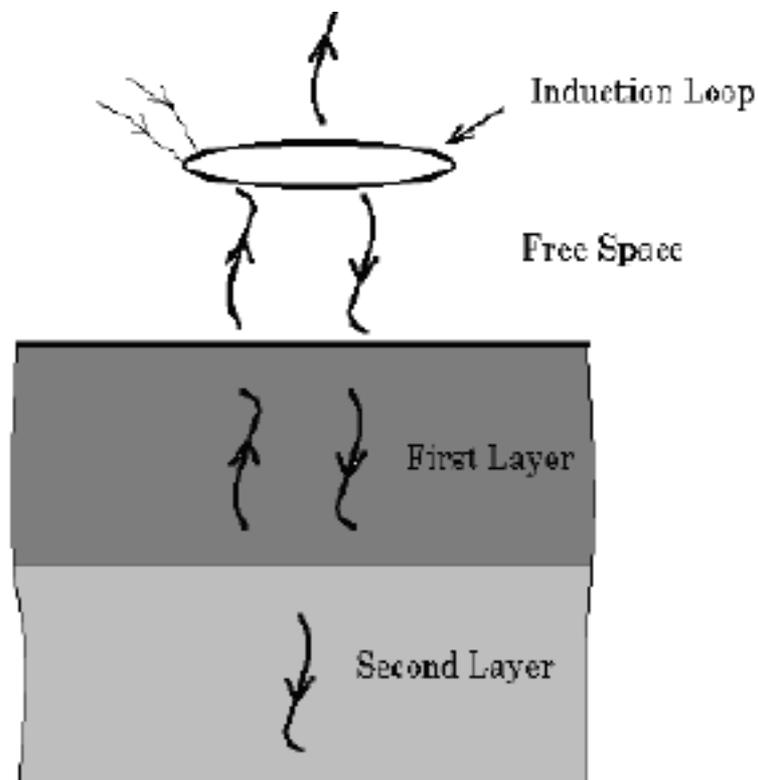


*RF & Spectrometer*  
*Ports*  
Casting Analysis Corporation/JLab

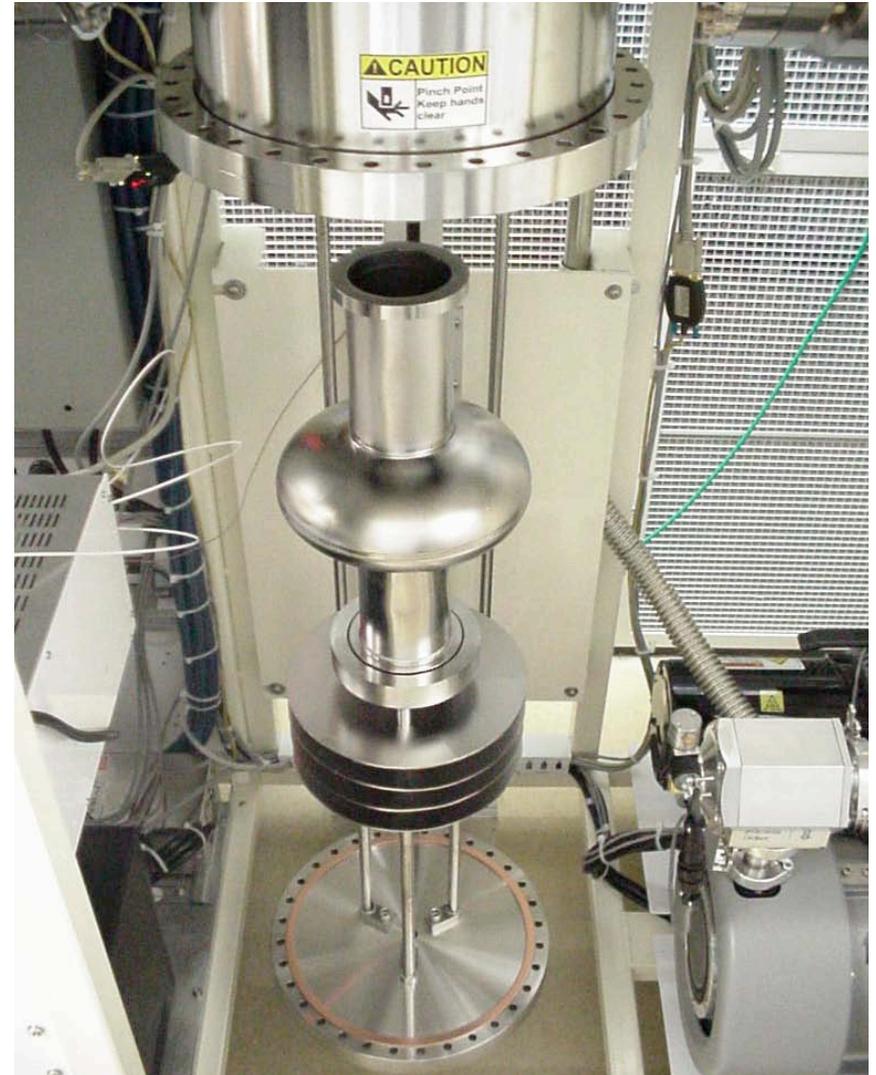
# Niobium-hydrogen measurement cells



# RF reflection measurement principle



# Clean UHV furnace - patents applied for

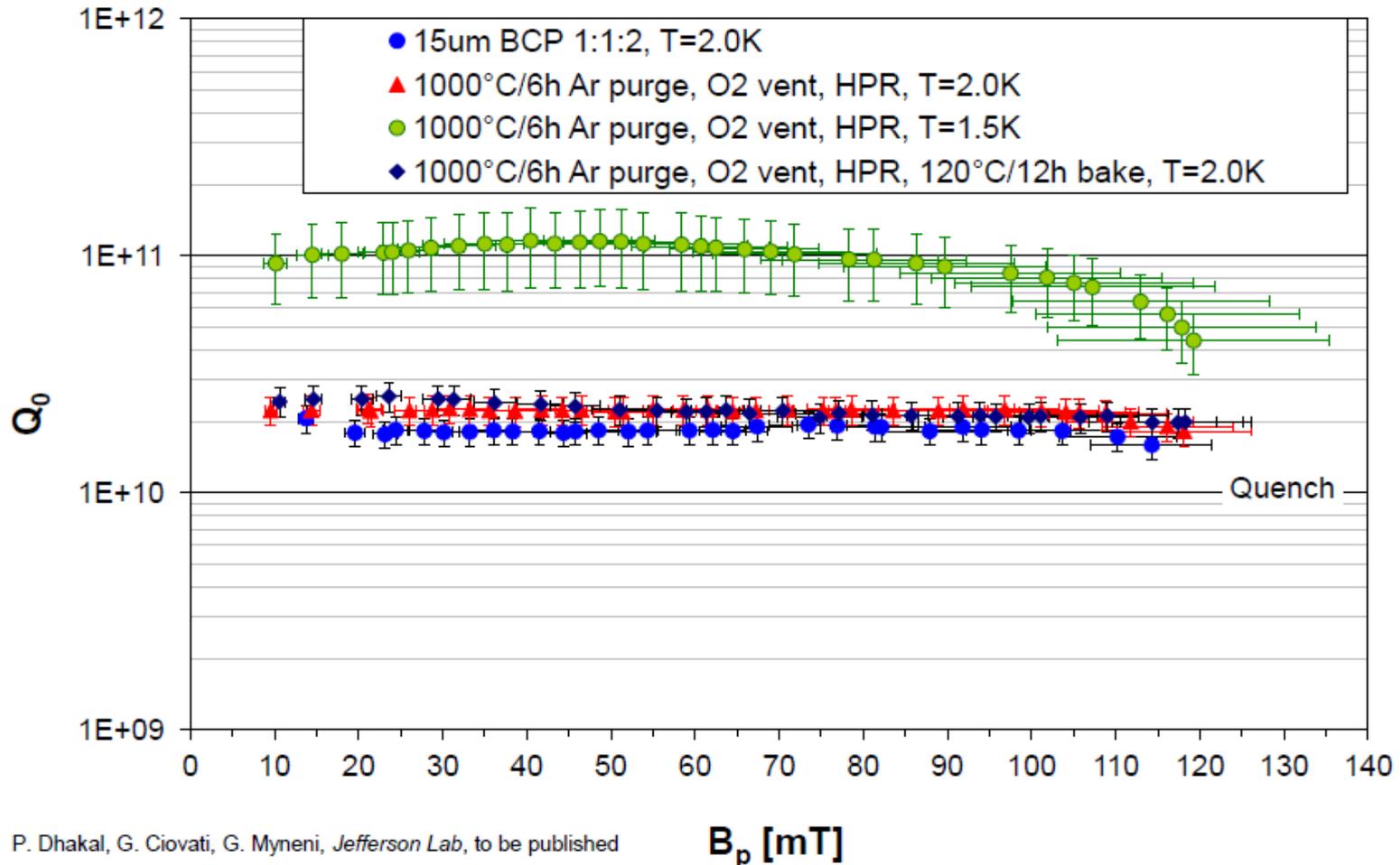


# Cavity material and preparation

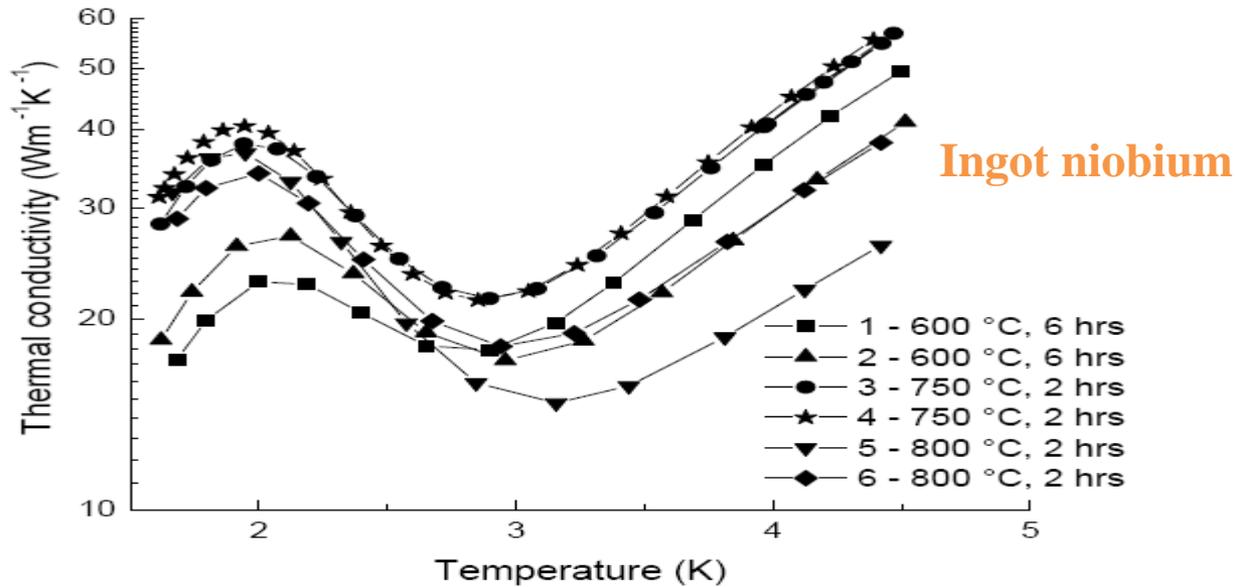
- CBMM ingot niobium, RRR  $\sim 200$  ( $>350$ ), Ta  $\sim 1350$  ( $<500$ ) wt ppm, inexpensive 50% to 60% less than conventional Nb
- Barrel polishing  $73\mu\text{m}$ , BCP  $65\mu\text{m}$  , a total of  $138\mu\text{m}$  removal and high pressure UHP water ( $\sim 200\mu\text{m}$  for fine grain)

# High and flat $Q_0$ – characteristic of ingot niobium

Large grain RRR ~ 200 Ta ~ 1375 CEBAF OC shape 1474 MHz cavity



# Tantalum and RRR have minimal influence on phonon peak



Specimen	Estimated RRR	Tantalum content (ppm) [3]	Heat Treatment	Titanium getter
1	191	1275	600 °C, 6 hrs	No
2	131	668	600 °C, 6 hrs	No
3	190	756	750 °C, 2 hrs	Yes
4	196	756	750 °C, 2 hrs	Yes
5	104	1322	800 °C, 2 hrs	No
6	143	523	800 °C, 2 hrs	No

MSU

# Cryogenic Refrigeration Cost Reduction with improved $Q_0$ (~factor of 3) CW SRF Cavities

- A 10 kW 2 K refrigerator costs ~ 100 M\$
- A factor of 3 improvement in  $Q_0$  will lower this to ~ 45 M\$
- The power consumption and hence the operating costs will be reduced by a third

# SSTIN10

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Myneni  
Ciovati  
Stuart

10<sup>7</sup> cm  
OH  
H  
International Symposium On  
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Jefferson Lab



Symposium on the Superconducting  
Science & Technology of Ingot Niobium

# Symposium on the Superconducting Science & Technology of Ingot Niobium



Jefferson Lab • Newport News, Virginia, USA

September 22-24, 2010

## Editors:

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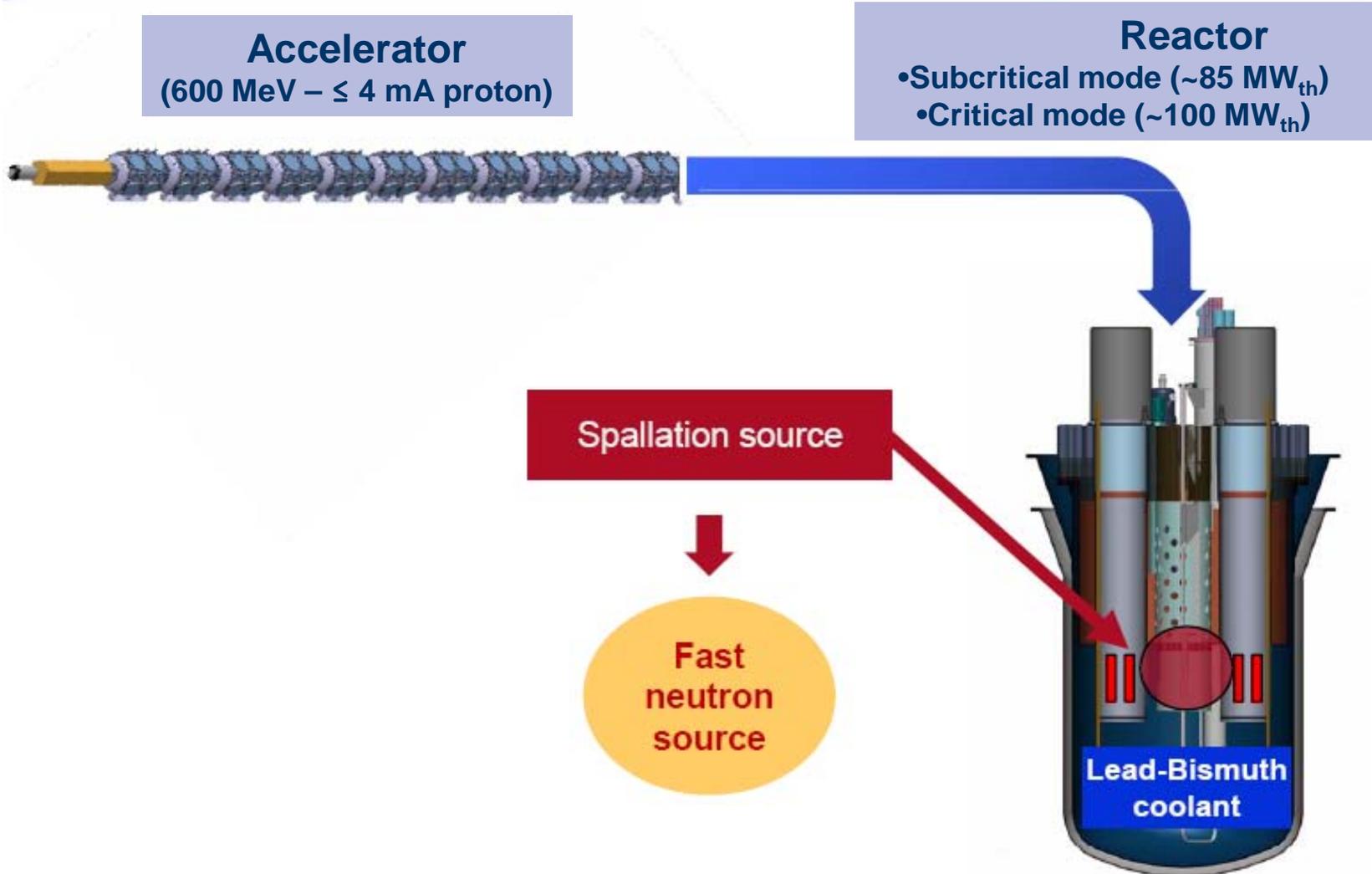
# Ingot niobium CW Applications

- Accelerator Driven Systems
- Energy Recovery Linacs for future light sources
- Compact Electron Linacs for Industrial and Medical Applications

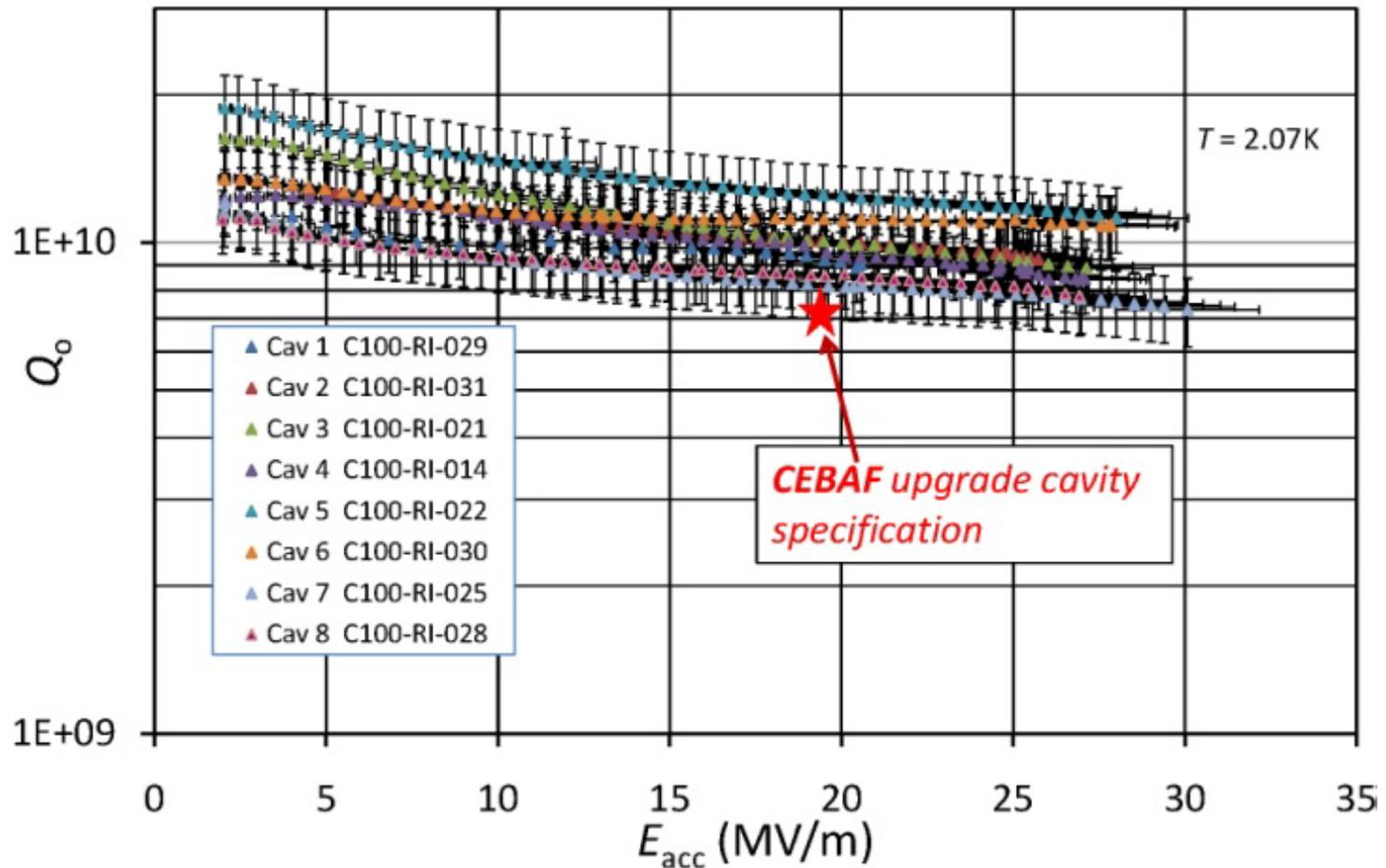
# Accelerator Driven Systems

- **Nuclear Waste Transmutation**
- **New nuclear fuel cycle material studies**
- **Energy Sustainability**
- **Carbon foot print reduction**

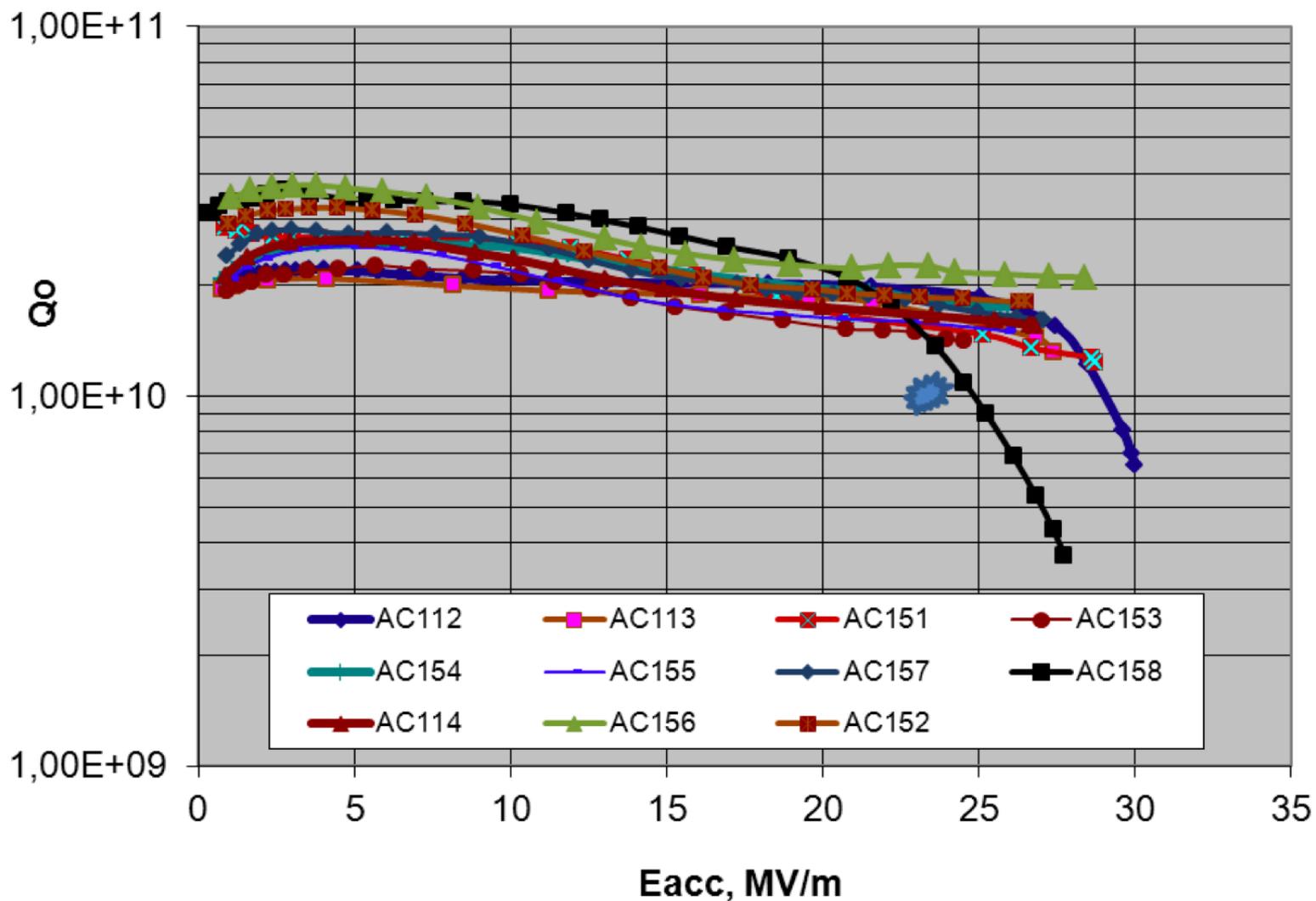
# MYRRHA Concept



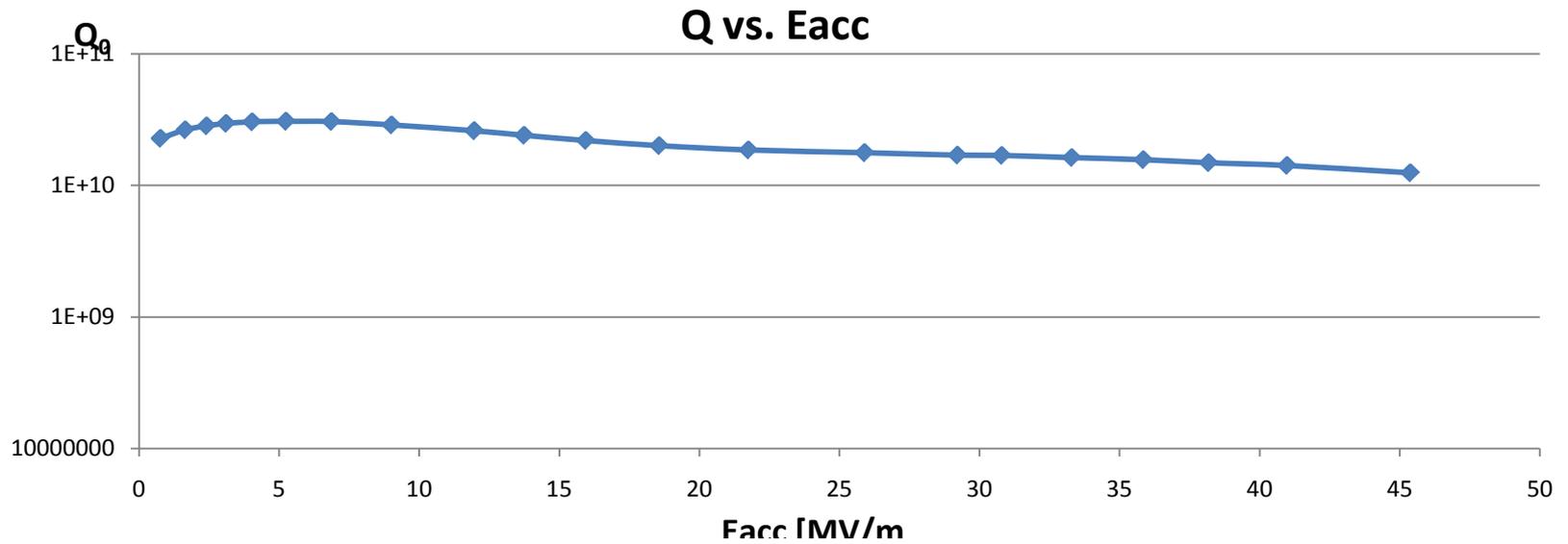
# CEBAF Upgrade cavities performance with the state of the art processes (EP)



# First Saclay built XFEL cryomodule will have all the ingot Nb cavities developed at DESY (Just BCP)



# DESY 9 Cell ingot XFEL cavity sets the world record



Waldemar Singer

# Summary

- \* Ingot niobium with high tantalum content meets CW SRF applications with minimum processing, lower cost and enhanced performance**
- \* Let us jointly work in implementing this cutting edge technology for energy sustainability at reduced carbon foot print**

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